

# Investigating the oxygen depletion effect in FLASH radiotherapy using GPU-based fast Monte Carlo simulations

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#### INTRODUCTION

- FLASH radiotherapy (RT), which delivers ionizing radiation at an ultrahigh dose rate (> 40Gy/s), is gaining increasing attention. It was found a better normal tissue sparing and a same or better tumor control probability, compared to the conventional RT at the same dose level.
- Oxygen depletion effect is one of the most popular hypothesis to explain FLASH RT effect. However, it remains controversial. It is almost impossible to detect this effect experimentally limited by the current technology, while results from phenomenological modeling is vulnerable with the empirical parameters selected.
- In principle, a step-by-step simulation via the Monte Carlo (MC) technique can be a good way to quantify the oxygen depletion effect in FLASH RT. However, it can be extremely time-consuming to simulate the oxygen chemical reaction with the CPU-based MC packages. Hence, almost no CPU-packages support this simulation functions.
- Recently, we can realize the step-by-step simulation of the oxygen depletion effect within our open source, GPU-accelerated MC package, gMicroMC [1].

# **AIM**

· In this work, we present our progress on applying the step-by-step oxygen depletion simulation via gMicroMC to quantify its impact in FLASH RT.

# **METHOD**

- · The details of implementing the step-by-step simulation of oxygen depletion effect in gMicroMC were given in another Blue Ribbon ePoster No. 53114. Table 1. Summary of chemical reactions.
- In this study, w used the chemic rates Plante's paper [2] with the channe also listed in Table Totally, 21 reactio were simulated.

we cal	Original reactions in gMicroMC	New reactions with oxygen
els om 2], els 1.	$\begin{array}{ll} e_h + e_h & \rightarrow 2OH^- + H_2 \\ e_h + OH^- & \rightarrow OH^- \\ e_h + H^+ & \rightarrow OH^- + H_2 \\ e_h + H^+ & \rightarrow H^- \\ e_h + H_2O_2 \rightarrow OH^- + OH^- \\ OH^- + OH^- \rightarrow H_2O_2 \\ OH^- + H^- & \rightarrow H_2O \\ H^- + H^- & \rightarrow H_2 \\ H^+ + OH^- & \rightarrow H_2O \\ H_2 + OH^- & \rightarrow H^- \end{array}$	$\begin{array}{cccc} e_h + O_2 & \rightarrow & O_2^- \\ e_h + HO_2 & \rightarrow & HO_2^- \\ e_h + O_2^- & \rightarrow & 2OH^- + H_2O_2 \\ OH^- + HO_2^- & \rightarrow & O_2 + OH^- \\ OH^- + HO_2^- & \rightarrow & HO_2^- + OH^- \\ OH^- + HO_2^- & \rightarrow & HO_2^- + OH^- \\ H^- + O_2^- & \rightarrow & H_2O_2 \\ H^- + O_2^- & \rightarrow & HO_2^- \\ H^+ + O_2^- & \rightarrow & HO_2^- \\ H^+ + HO_2^- & \rightarrow & H_2O_2 \\ \end{array}$

#### ➤ Simulation setup

• Spatial distribution (Figure 1)

We randomly sample electrons with energy  $E_k$  in a sphere with radius R, which is equal to the radius of Region of Interest (ROI) r plus the maximum travel distance of the electron. Only radicals produced inside ROI will be considered in the chemical stage simulation. Oxygen is uniformly distributed inside a cube with a dimension of d = 2(r +0.3) µm. Oxygen regeneration is ignored because of the very short pulse (1  $\mu$ s). Oxygen numbers are determined from the oxygen concentration  $C_{O_2}$  as  $N_{O_2} = C_{O_2} * 760 \text{ mmHg} * 1.26 *$ 

 $\frac{\mu M}{mmHg}*d^3*N_A$ , where  $N_A$  is the Avogadro constant. Oxygen concentration levels of 0.01%, 0.1%, 0.5%, 1%, 3%, 9%, and

Figure 1. Illustration of the simulation geometry. Blue area (r): ROI. Green area (R): source particles to be sampled in. Orange area (d): oxygen to be sampled in.

Temporal distribution

Temporally, a beam pulse width of  $1 \mu s$  is used to sample the electrons. The initial electrons are uniformly sampled within the pulse until the instantaneous dose rate  $D_i$  in the beam pulse reaches predefined values of  $10^6 Gy/s$ ,  $10^7 Gy/s$ and  $10^8 Gy/s$ .

· Energy and others

Specifically in this work,  $E_k = 4.5 \text{ keV}$  and 0.3 keV are considered. The 4.5 keV is used to represent radiation cases with primary particles being electrons and photons, while the 0.3 keV is intended to show cases of primary proton and heavy ions, the energies of secondary electrons of which are usually of hundreds of eVs. The cutoff energy for the physical stage is 7.5 eV. Default values in gMicroMC for physicochemical stage are used [1]. The duration for chemical stage is 2  $\mu s$  to ensure full reaction between chemical species.

Evaluation metrics

Time-dependent yields of radicals, oxygen consumption rate (OCR), residual  $C_{O_2}$  were used for the quantification for different instantaneous dose rate  $D_i$ .  $OCR = \Delta G(O_2) * \rho/N_A$ with  $\rho$  the density of the medium and  $\Delta G = \Delta N(O_2)/E_r$ , where  $E_r$  is the total deposited energy in the ROI and  $\Delta N(O_2)$ is the change of the number of oxygen molecules.

#### RESULTS

#### ☐ Time-dependent yields of radicals (Figure 2)

- At both oxygen levels, yields of superoxide radicals are similar between that with a dose rate of  $10^6$  Gy/s and  $10^7 \, Gy/s$  ((a1) vs. (b1), and (a2) vs. (b2)). It then drops quickly at the dose rate of  $10^8$  Gy/s, compared to the other lower dose rates. It indicates that competition between different chemical reactions can happen when dose rate is too high to cause overlap between different
- A higher initial oxygen ( $C_{0_2} = 3\%$ ) leads to a higher yield of superoxide radicals compared to the lower concentration ( $C_{O_2} = 0.1\%$ ), indicating more oxygen consumption.

#### ☐ Calculated OCR (Figure 3)

☐ Residual percentage (Figure 4)

and  $C_{O_2} = 3\%$ , respectively.

- OCR drops with increasing dose rates at the same oxygen concentration level;
- OCR drops with decreasing initial oxygen concentration levels when the dose rate is the same.
- varying OCR, while it can be fully depleted if a constant OCR at  $C_{O_2} = 21\%$  is assumed.

From left column of Figure 4, a higher dose rate can lead

From right column of Figure 4, if oxygen regeneration is

consumption. This can be explained by the chemical

enhancement ratio (OER),  $OER = 1 + 1.63 \times (1 - 1.00)$ 

reaction competition effect observed in Figure 2.

According to the empirical formula for oxygen

not considered, a lower dose rate leads to a higher oxygen

 $e^{-0.26*C_{02}*760}$ ) [3], we can estimate that OER drops from

1.3 to 1.2 for  $C_{O_2} = 0.1\%$  (hypoxia, 0.1% to 0.06%) and

from 2.63 to 2.61  $C_{O_2} = 3\%$  (physoxia, 3% to 2.4%) for

in Figure 4(b<sub>2</sub>) after 30 Gy radiation. The percentage

change of OER is only 7.7% and 0.8% for  $C_{0a} = 0.1\%$ 

to a higher oxygen consumption since it corresponds a

higher initial distributions of chemical radicals;

# Oxygen can not be fully depleted by radiolysis with

column) after one pulse for (a) 4.5 keV (b) 0.3 keV under different initial oxygen

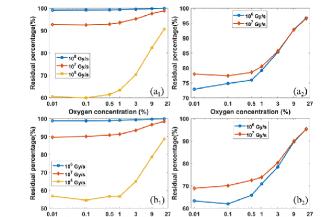


Figure 4. Calculated residual percentage of  $C_{O_2}$  under different dose rates for (a) 4.5 keV (b) 0.3 keV. The left and tight column represent the results for a pulse length of 1  $\mu s$  and an accumulated dose of 30 Gy without considering oxygen regeneration, respectively

# **CONCLUSIONS**

With the help of GPU in accelerating the chemical stage, we were able to simulate the water radiolysis with the presence of oxygen to study the oxygen depletion effect in FLASH RT. Our simulation showed that the OCR should decrease with increasing dose rate and decreasing initial oxygen concentration, leading to the inability of full oxygen depletion by radiolysis at all tested oxygen concentration level. Without considering oxygen regeneration between pulse, the decreasing percentage of OER brought by the oxygen consumption is only 7.7% and 0.8% for initial hypoxia and physoxia condition after 30 Gy radiation.

We will continue to study the oxygen effect on FLASH RT by considering the oxygen regeneration, DNA repair, etc. so that other factors like the time interval between pulses can be taken into consideration.



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### REFERENCES

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## **CONTACT INFORMATION**

For any interest in the package or collaboration, please feel free to contact Xun Jia (xun.jia@utsouthwestern.edu) and Yujie Chi (yujie.chi@uta.edu).

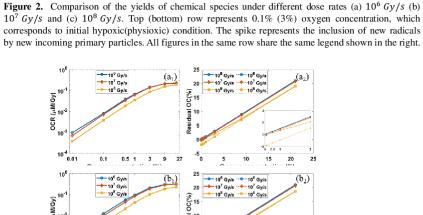


Figure 3. Calculated OCR (left column) and residual oxygen concentration (right